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Supplementary Data

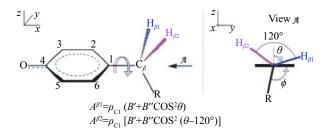
# Tyrosyl radical in haemoglobin and haptoglobin-haemoglobin complex: how does haptoglobin make haemoglobin less toxic?

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## **Updated TRSSA - what is new?**

Tyrosyl Radical Spectra Simulation Algorithm (TRSSA)<sup>[1]</sup> uses two input parameters, spin density on atom C1  $\rho_{C1}$  and phenoxyl ring rotation angle  $\theta$  (*Supplementary Fig. 1*), to calculate all parameters needed for simulation of a Tyr radical EPR spectrum. Euler angles for the methylene  $\beta$ -protons hyperfine interaction tensors were set to zeros in the original TRSSA. We now used Density Function Theory (DFT) calculations to determine the Euler angles for the protons as functions of the ring rotation angle  $\theta$ .

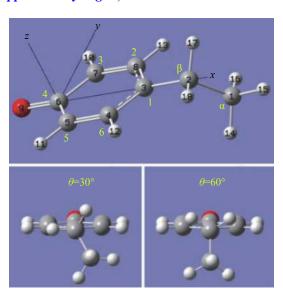


Supplementary Fig. 1 Tyrosyl radical, the Cartesian coordinate system (right-handed) aligned with the principal g -factor value directions and the definition of the phenoxyl ring rotation angle. The hyperfine interaction constants A for the methylene protons  $\beta 1$  and  $\beta 2$  are proportional to the spin density  $\rho_{C1}$  on atom C1 and depend on  $\theta$  according to the McConnell relationship (B' and B'' are constants) [2]. Angle  $\varphi$  is an alternative definition of the ring rotation, more practical for crystal structure analysis, when H-atoms are not visible.

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# DFT calculation of Euler angles for a model Tyr radical

DFT was used to determine the Euler angles for the  $\beta$ -protons in a model Tyr radical for two rotational conformations of the ring, with  $\theta$ =30° and  $\theta$ =60° (Supplementary Fig. 2).



Supplementary Fig. 2 The Tyr model and two ring conformations analysed. At the top: the model Tyr radical used in the DFT calculations. Atom numbers assigned by Gaussian are indicated (in a smaller font on the atoms) along with traditional atom nomenclature in a tyrosine (in a bigger font by the atoms). At the bottom: the model at two different conformations, with the ring rotation angles  $\theta$ =30° and  $\theta$ =60°, shown for view A as define in Supplementary Fig. 1.

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The authors reported no conflict of interests.

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Gaussian03 package<sup>[3]</sup> was used for both the optimisations of the structures and the calculation of the EPR parameters, both at the B3LYP/6-31G\* level of theory. During the optimization, the atom coordinates that define the rings rotation angles were kept frozen. After optimization, the Cartesian axes, changed by the optimisation process, have been reassigned as shown in Supplementary Fig. 2. This was done by using ArgusLab 4.0.1<sup>[4]</sup> as follows. Ox was first set through C6-C2 (Gaussian nomenclature); Oy was then linked to C6-C7, that moved Ox away from the C6-C2 direction but set Oz perpendicular to the ring plane. Now Ox was re-aligned to C6-C2 which moved Oy away from C7 but did keep it within the ring plane. Single point calculations were then performed with the NoSymm keyword to prevent molecule re-orientation. The Gaussian output files for the two single point calculations are available in the Supplementary theta 30 degrees.out and theta 60 degrees.out (all the supplementary files are available on: https://ldrv.ms/u/s!ArZmAEA-E2XThrNQz79qU XuUqIrL-A?e=4fNPKw).

Euler angles  $\Phi 1$ ,  $\Phi 2$  and  $\Phi 3$  for a proton's A-tensor were calculated from the projections of three unity vectors, aligned with the directions aa, bb and cc of the Anisotropic Spin Dipole Coupling components, onto Cartesian axes x, y and z. These projections are available in the Gaussian output, and the A-values

and Euler angles spreadsheet of Supplementary Euler angles.xlsx was used perform the calculations. This Excel file was adapted from the file published the Electronic Supplementary Information to our work published previously[5]. Attention has been paid to the choice of left and right Cartesian coordinate systems and to different definition of the coupling constants in Gaussian ( $B_{aa}$  smallest,  $B_{bb}$  - intermediate,  $B_{cc}$  - highest) and in SimPow6 ( $|A_{aa}|$  - smallest,  $|A_{bb}|$  - intermediate,  $|A_{cc}|$  highest-by absolute value). The three Euler angles are defined as follows:

•First rotation  $\Phi I$  is around z when z is directed towards the observer, to bring  $A_{cc}$  to the zx plane.

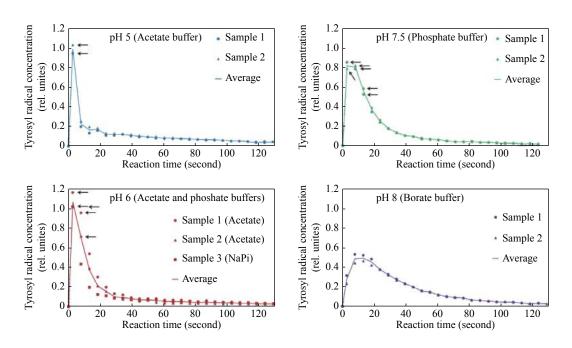
•Second rotation  $\Phi 2$  is around y when y is directed away from the observer, to align  $A_{cc}$  with x.

•Third rotation  $\Phi 3$  is around x when x is directed towards the observer;  $\Phi 3$  is positive when measured counter clock-wise from positive direction of y.

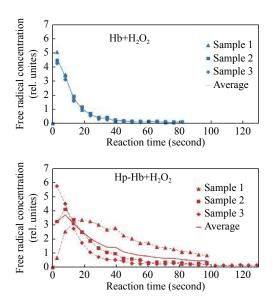
The formulae for the calculation of the Euler angles can be viewed by selecting appropriate cells on *Avalues\_and\_Euler\_angles* spreadsheet and are derived on the *Euler\_angles\_explained* spreadsheet of *Supplementary Euler\_angles.xlsx*.

## Euler angles for a rotational orientation of the ring between 30° and 60°

In TRSSA2, the Euler angles for a specific ring



Supplementary Fig. 3 Kinetic dependences of the tyrosyl radical in the individual samples of 340 μmol/L metHb mixed with a 4-fold excess of H<sub>2</sub>O<sub>2</sub> (1.36 mmol/L, both concentrations are final), at the range of pH values of pH 5-pH 8 maintained by different buffers. Averaged data (lines) for the four pH values are shown on one graph in Fig. 3 as data points. The thirteen EPR spectra corresponding to the data points marked with arrows have been averaged to produce the spectrum in inset in Fig. 3. The nine first spectra of Sample 1 at pH 7.5 (up to 45 seconds) are shown in Fig. 1.



Supplementary Fig. 4 Kinetic dependences of the free radical in three similar experiments of mixing 130 μmol/L Hb with 1.558 mmol/L H<sub>2</sub>O<sub>2</sub> and in three similar experiments of mixing 130 μmol/L Hp-Hb with 1.558 mmol/L H<sub>2</sub>O<sub>2</sub>, at pH 7.4. The averaged data (solid lines) are plotted on one graph in Fig. 5. Note that while the kinetics of the free radical in unbound Hb are very close in the three similar experiments, this is not so for the Hp-Hb complex: the three similar Hp-Hb+H<sub>2</sub>O<sub>2</sub> experiments exhibit rather different kinetics. While the cause of this variability should be investigated in the future works, such kinetic behavior seems to be the reason for rather different relative increase of the free radical yield in the Hp-Hb complex, as compared to unbound Hb, observed before in the frozen samples<sup>[6–7]</sup>.

rotational conformation  $\theta$ , which is within the interval of  $30^{\circ} \le \theta \le 60^{\circ}$ , were calculated as linear interpolation of the DFT determined values for  $\theta$ =30° and  $\theta$ =60°. This interval of  $\theta$  was selected because the original TRSSA has yielded this angle to be 53° in the human Hb radical<sup>[1]</sup> and because this angle was estimated as 47° in the Hp-Hb complex<sup>[6]</sup>. Thus the angle in both free and Hp-bound Hb was expected to be within the 30°-60° interval.

## **EPR** spectra simulation parameters

The parameters used to simulate the free radical EPR spectra of Hb and Hp-Hb complex were generated by TRSSA2 and are summarized in *Supplementary Table 1* and *Table 2*, respectively. The SimPow6 input data files (*Supplementary d\_Hb.dat* and *d\_Hb\_Hp.dat*) were based on these tables.

	$g_{ m y}$	$g_{\rm z}$			
2.00707	2.00421	2.00221			
Hyperfine coupling matrix principal values			Euler angle $\alpha$	Euler angle β	Euler angle γ
$A_{cc}^{\beta 1}$ , MHz	$A_{bb}^{\beta 1}$ , MHz	$A_{aa}^{\beta 1}$ , MHz	$-\Phi_1$ <sup>1</sup> , degree	$-\Phi_2^{-1}$ , degree	$-\Phi_3^1$ , degree
24.19	20.70	20.70	-23.1	-11.5	-39.6
$A_{cc}^{\beta 2}$ , MHz	$A_{bb}^{\beta 2}$ , MHz	$A_{aa}^{\beta 2}$ , MHz	$-\Phi_1^2$ , degree	$-\Phi_2^2$ , degree	$-\Phi_{3}^{2}$ , degree
12.28	7.68	7.68	25.4	-6.7	-41.8
$A_{cc}^{C3}$ , MHz	$A_{bb}^{C3}$ , MHz	$A_{aa}^{C3}$ , MHz	$-\Phi_1^{C3}$ , degree	$-\Phi_2^{C3}$ , degree	$-\Phi_3^{C3}$ , degree
-25.9	-8.1	-20.5	23.0	0.0	0.0
$A_{cc}^{C5}$ , MHz	$A_{bb}^{C5}$ , MHz	$A_{aa}^{C5}$ , MHz	$-\Phi_{\mathrm{l}}^{\mathrm{C5}}$ , degree	$-\Phi_2^{C5}$ , degree	$-\Phi_3^{C5}$ , degree
-25.9	-8.1	-20.5	-23.0	0.0	0.0
$A_{cc}^{C2}$ , MHz	$A_{bb}^{C2}$ , MHz	$A_{aa}^{C2}$ , MHz	$-\Phi_1^{C2}$ , degree	$-\Phi_2^{C2}$ , degree	$-\Phi_3^{C2}$ , degree
7.5	5.0	1.5	40.0	0.0	0.0
$A_{cc}^{C6}$ , MHz	$A_{bb}^{C6}$ , MHz	$A_{aa}^{C6}$ , MHz	$-\Phi_1^{C6}$ , degree	$-\Phi_2^{C6}$ , degree	$-\Phi_3^{C6}$ , degree
7.5	5.0	1.5	-40.0	0.0	0.0
$\Delta H_x$ , Gauss	$\Delta H_y$ , Gauss	$\Delta H_z$ , Gauss			

Supplementary Table 2 Parameters used to simulate the Hp-Hb free radical EPR spectrum (Fig. 6) – a TRSSA output for an input of  $\rho_{CI}$ =0.360 and  $\theta$ =54.3°

$g_{\mathrm{x}}$	$g_{ m y}$	$g_z$			
2.00868	2.00435	2.00215			
Hyperfine coupling matrix principal values		Euler angle $\alpha$	Euler angle β	Euler angle γ	
$A_{cc}^{\beta 1}$ , MHz	$A_{bb}^{\beta 1}$ , MHz	$A_{aa}^{\beta 1}$ , MHz	$-\Phi_1^1$ , degree	$-\Phi_2^1$ , degree	$-\Phi_3^1$ , degree
20.34	17.23	17.23	-23.4	-10.9	-38.6
$A_{cc}^{\beta 2}$ , MHz	$A_{bb}^{\beta 2}$ , MHz	$A_{aa}^{\beta 2}$ , MHz	$-\Phi_1^2$ , degree	$-\Phi_2^2$ , degree	$-\Phi_3^2$ , degree
11.71	7.77	7.77	25.4	-6.9	-43.9
$A_{cc}^{C3}$ , MHz	$A_{bb}^{C3}$ , MHz	$A_{aa}^{C3}$ , MHz	$-\Phi_1^{C3}$ , degree	$-\Phi_2^{C3}$ , degree	$-\Phi_3^{C3}$ , degree
-25.9	-8.1	-20.5	23.0	0.0	0.0
$A_{cc}^{C5}$ , MHz	$A_{bb}^{C5}$ , MHz	$A_{aa}^{C5}$ , MHz	$-\Phi_1^{C5}$ , degree	$-\Phi_2^{C5}$ , degree	$-\Phi_3^{C5}$ , degree
-25.9	-8.1	-20.5	-23.0	0.0	0.0
$A_{cc}^{C2}$ , MHz	$A_{bb}^{C2}$ , MHz	$A_{aa}^{C2}$ , MHz	$-\Phi_1^{C2}$ , degree	$-\Phi_2^{\text{C2}}$ , degree	$-\Phi_3^{C2}$ , degree
7.5	5.0	1.5	40.0	0.0	0.0
$A_{cc}^{C6}$ , MHz	$A_{bb}^{C6}$ , MHz	$A_{aa}^{C6}$ , MHz	$-\Phi_{ m l}^{ m C6}$ , degree	$-\Phi_2^{\text{C6}}$ , degree	$-\Phi_3^{C6}$ , degree
7.5	5.0	1.5	-40.0	0.0	0.0
$\Delta H_x$ , Gauss	$\Delta H_y$ , Gauss	$\Delta H_z$ , Gauss			
4.25	2.79	2.71			

Orange indicates radical parameters generated by the algorithm for an input of  $\rho_{C1}$  and  $\theta$ ; Light blue indicates conserved parameters (the same for all Tyr radicals).

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